

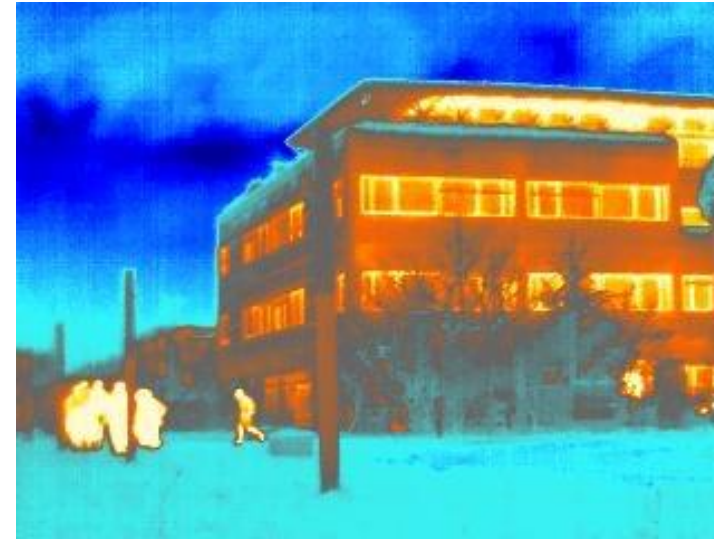


Photo: A. Black, UBC

07 Long-wave radiation.

Learning objectives

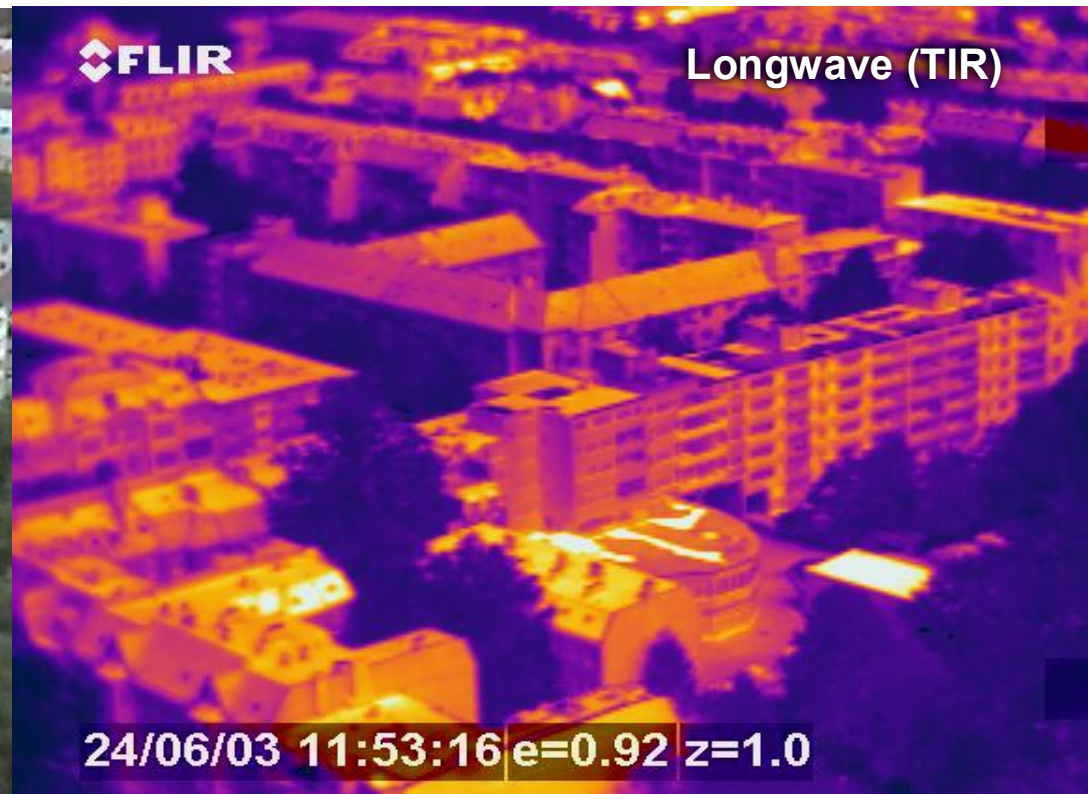
- Explain how radiation laws apply to long-wave radiative exchange as well.
- Describe how long-wave radiation can interact with surfaces.
- Explain how we can calculate longwave outgoing radiation, and how it relates to surface emissivity.



People in front of a heated building in winter as seen in the thermal infrared by a thermal camera (Source: A. Christen)

What is 'Longwave' radiation?

- Wavelength range: 3 μm to 100 μm
- Longwave = far-infrared = thermal infrared radiation (TIR)



Measuring long-wave radiation

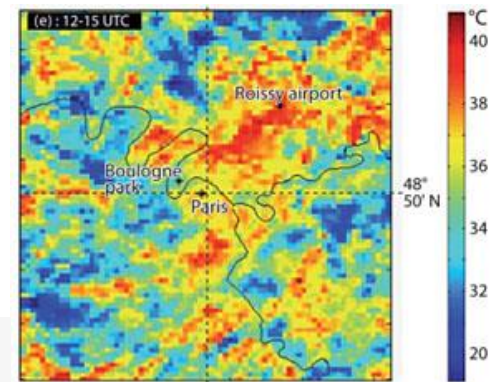
There is a range of instruments available that measure long-wave radiation received in a particular **field of view** (FOV), within a particular **band**, and/or with a particular spatial **resolution**.



Pyrometer



Thermal camera



Thermal satellite channel

Review: Stefan-Boltzmann law: grey body

Natural objects are not full radiators. The **emittance** from these objects (called grey bodies) is given by

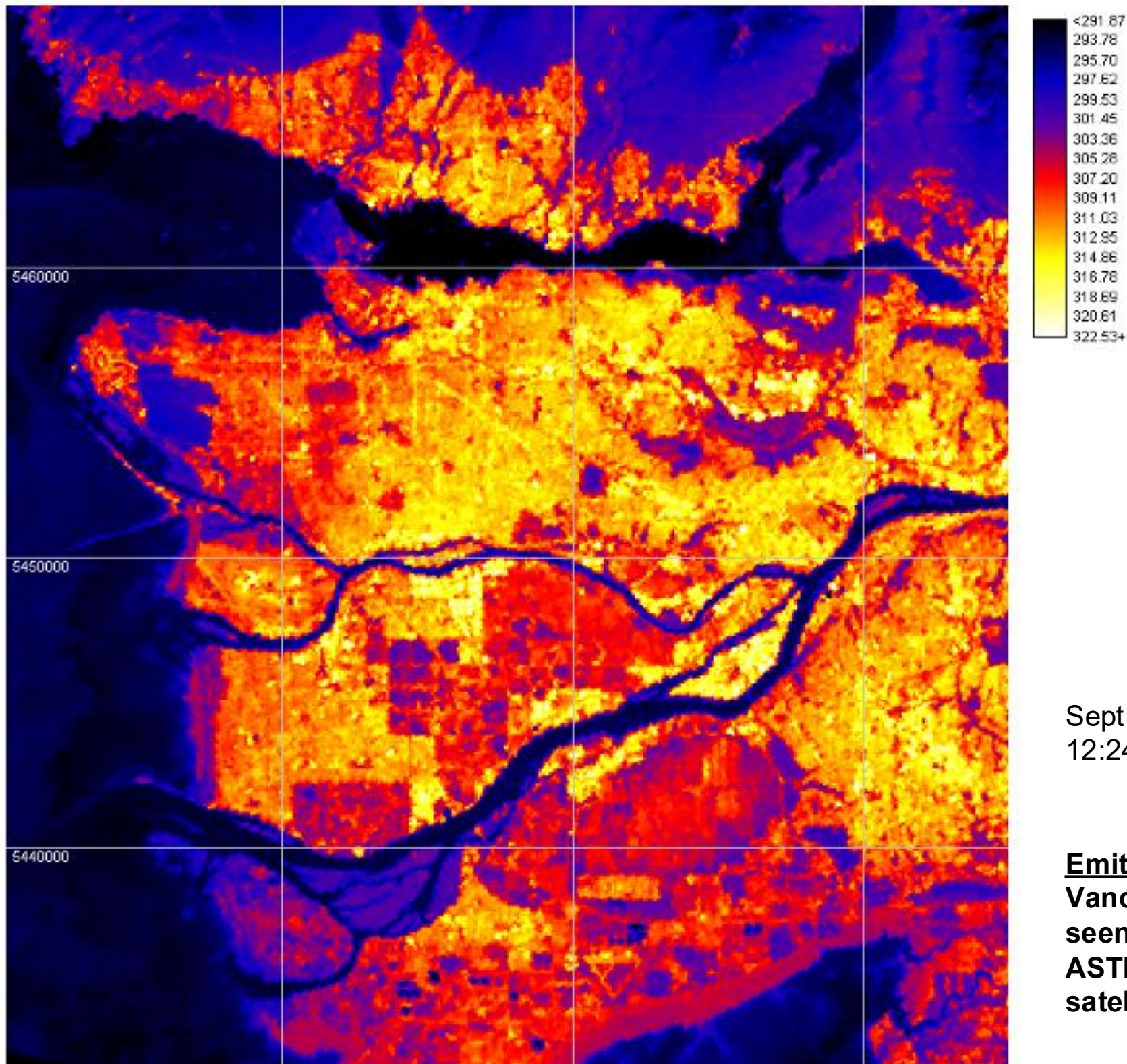
$$E_g = \varepsilon \sigma T^4 \quad \star$$

where ε is their **surface emissivity**. Emissivity is the ratio of the actual emission to that of a blackbody (i.e. $\varepsilon = 1.0$).

This law is the basis of remote sensing in the TIR incl. satellite sensors.

Surface	Emissivity ε^*
Soil	0.90 – 0.98
Grass	0.90 – 0.95
Crops	0.90 – 0.99
Forests	0.97 – 0.99
Water	0.92 – 0.97
Iron	0.13 – 0.28

* in the long-wave range of the spectrum (relevant for microclimates)

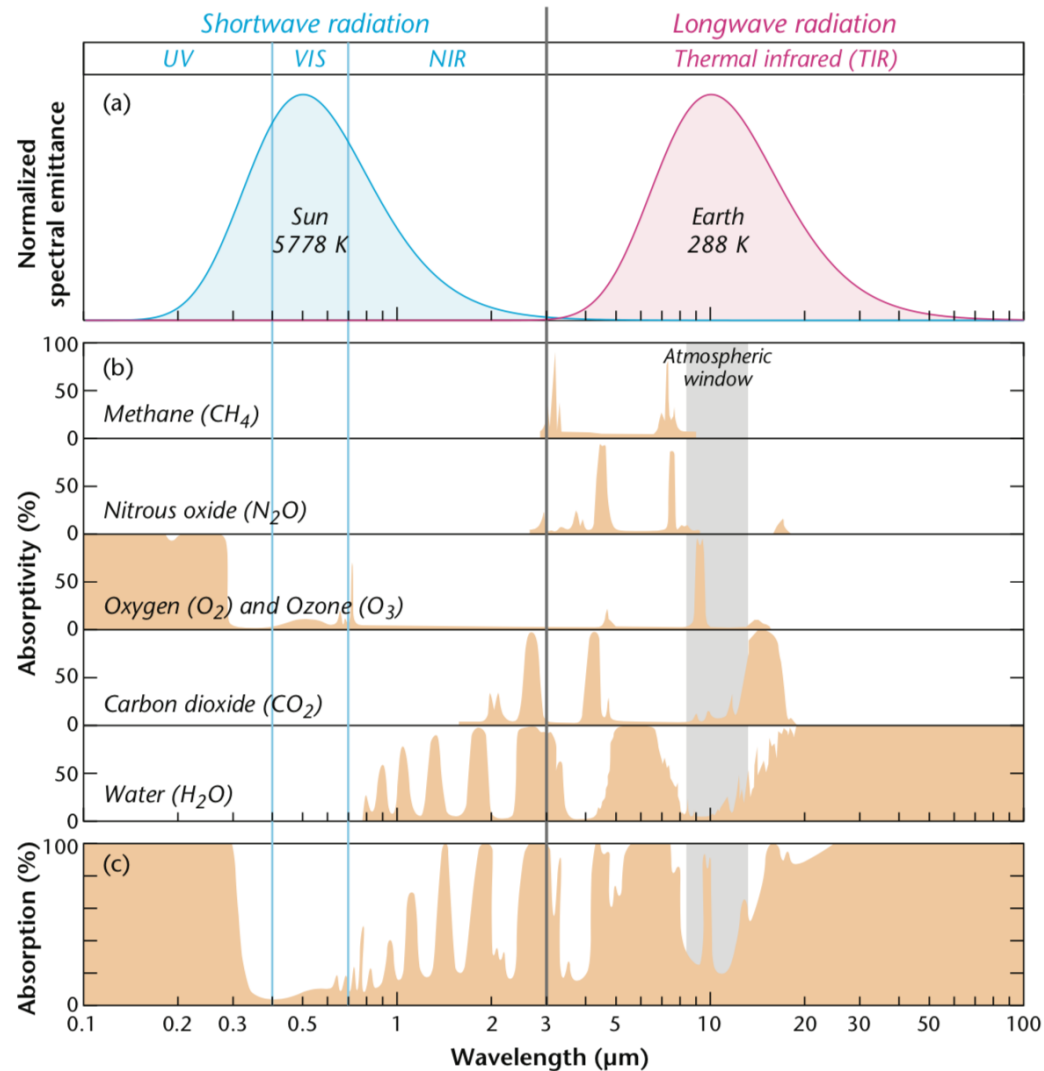


$W m^{-2}$

Sept 3 2010
12:24 PDT

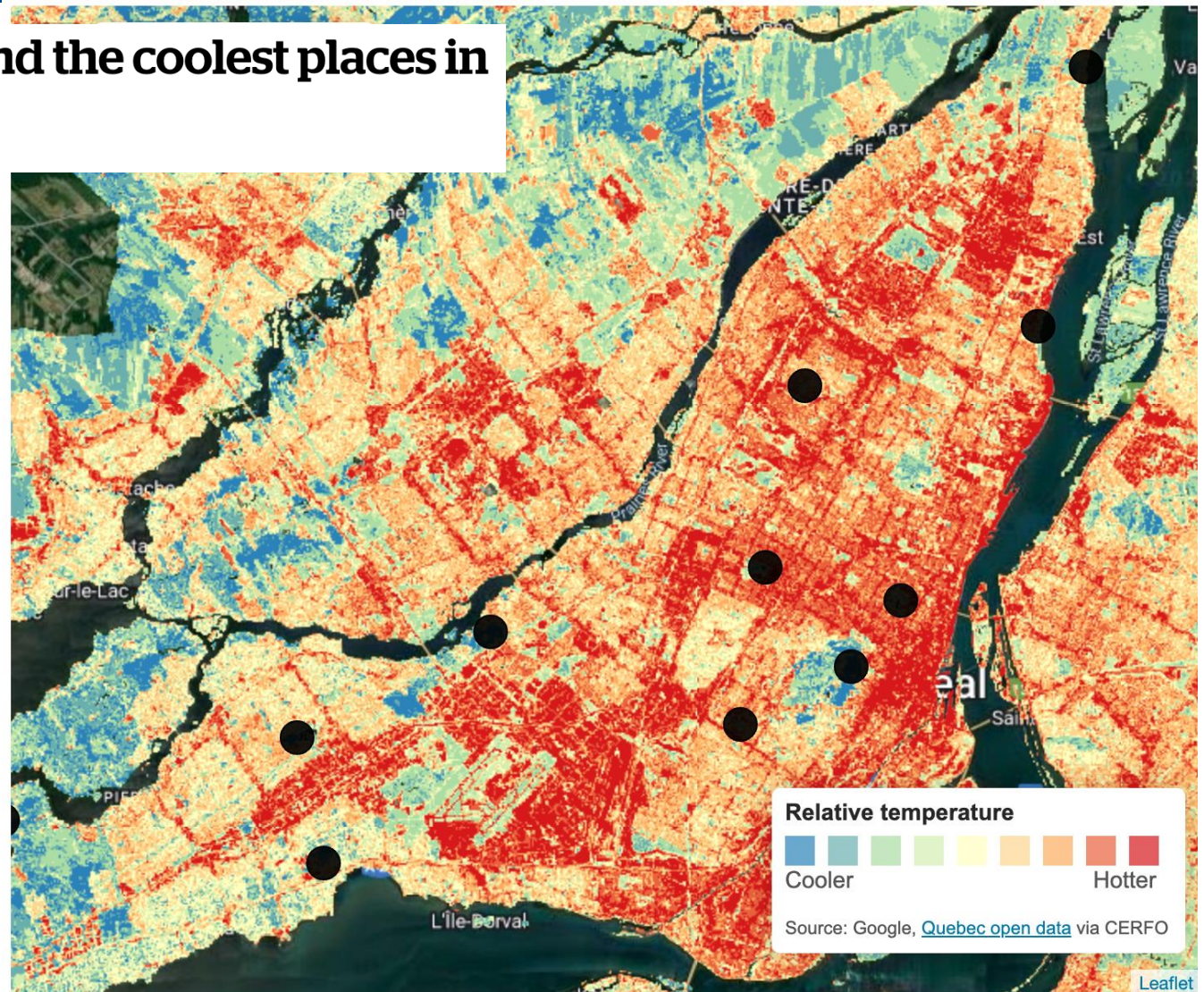
**Emittance of
Vancouver
seen from
ASTER
satellite**

Atmospheric window & remote sensing in the TIR



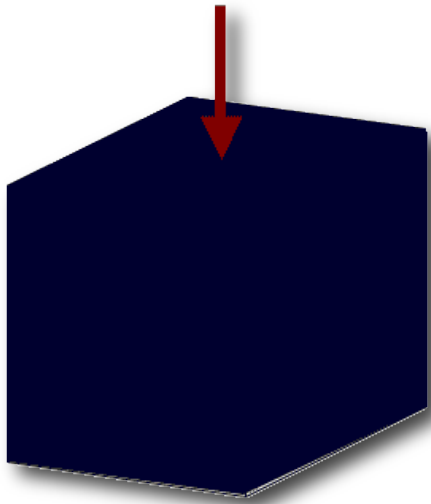
From emittance we can estimate surface temperature

Where are the warmest and the coolest places in Montreal?

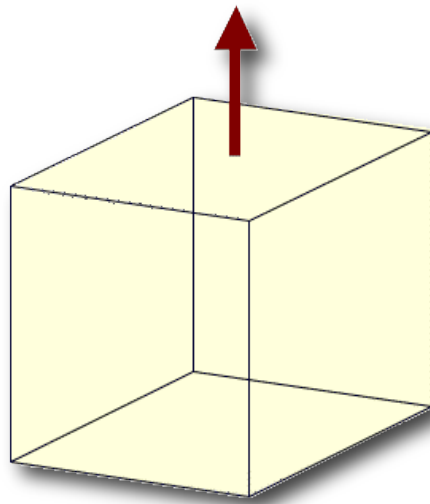


Review: Kirchhoff's law

Absorptivity



Emissivity



Assuming no transmission the absorptivity of a body (ζ_λ) equals its emissivity (ϵ_λ) at a given wavelength.

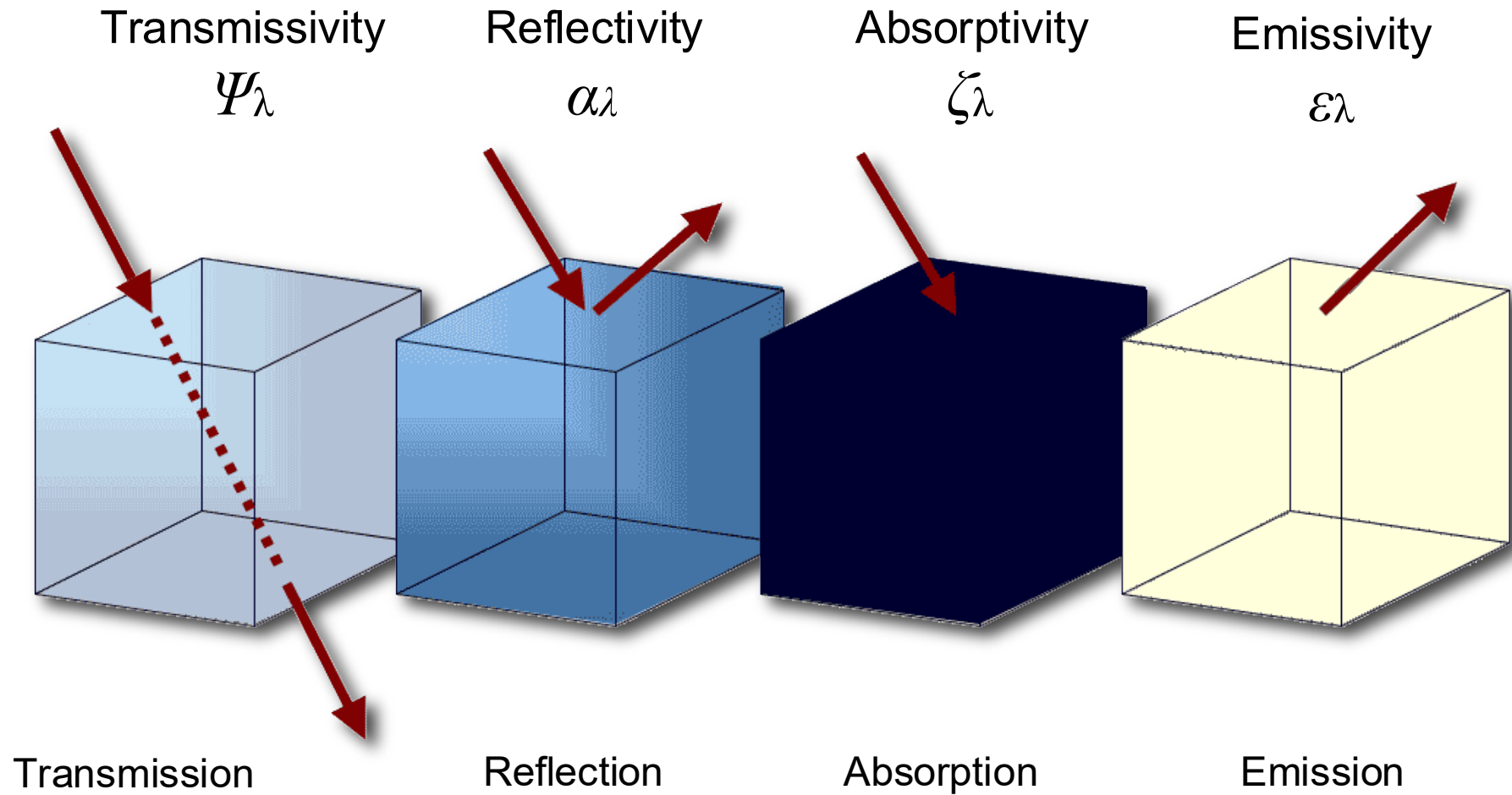
A good absorber is a good emitter

$$\zeta_\lambda = \epsilon_\lambda \quad \star$$

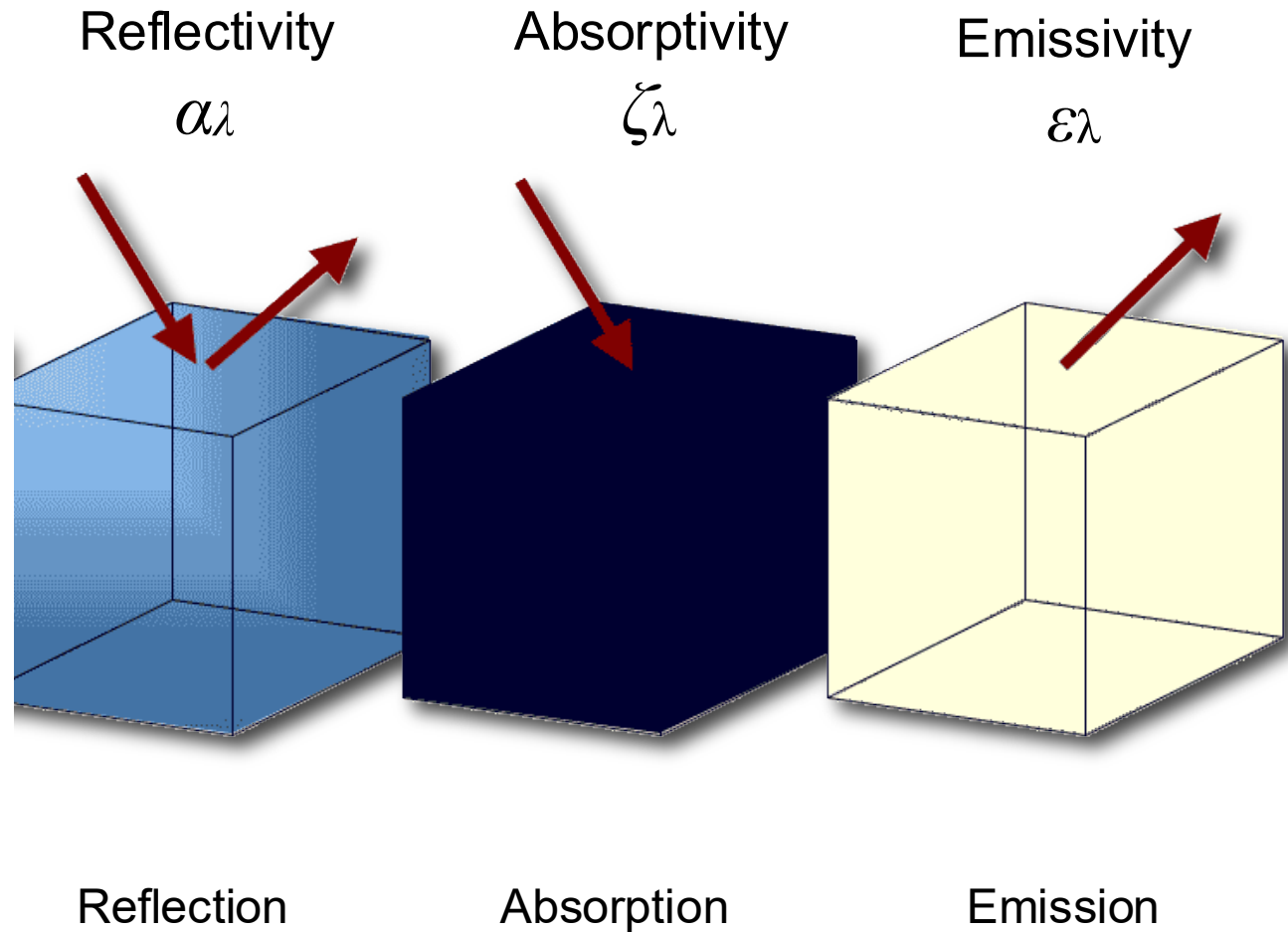
Kirchhoff's law of thermal radiation

- Kirchhoff's law only applies if the wavelength considered is the same – do not mix them together.
- Kirchhoff's law only has relevance to **long-wave exchange** in climatology.
- The law does not apply to fluorescent objects, which can absorb energy at a given a wavelength and release it at another one.

Mass-radiation interactions

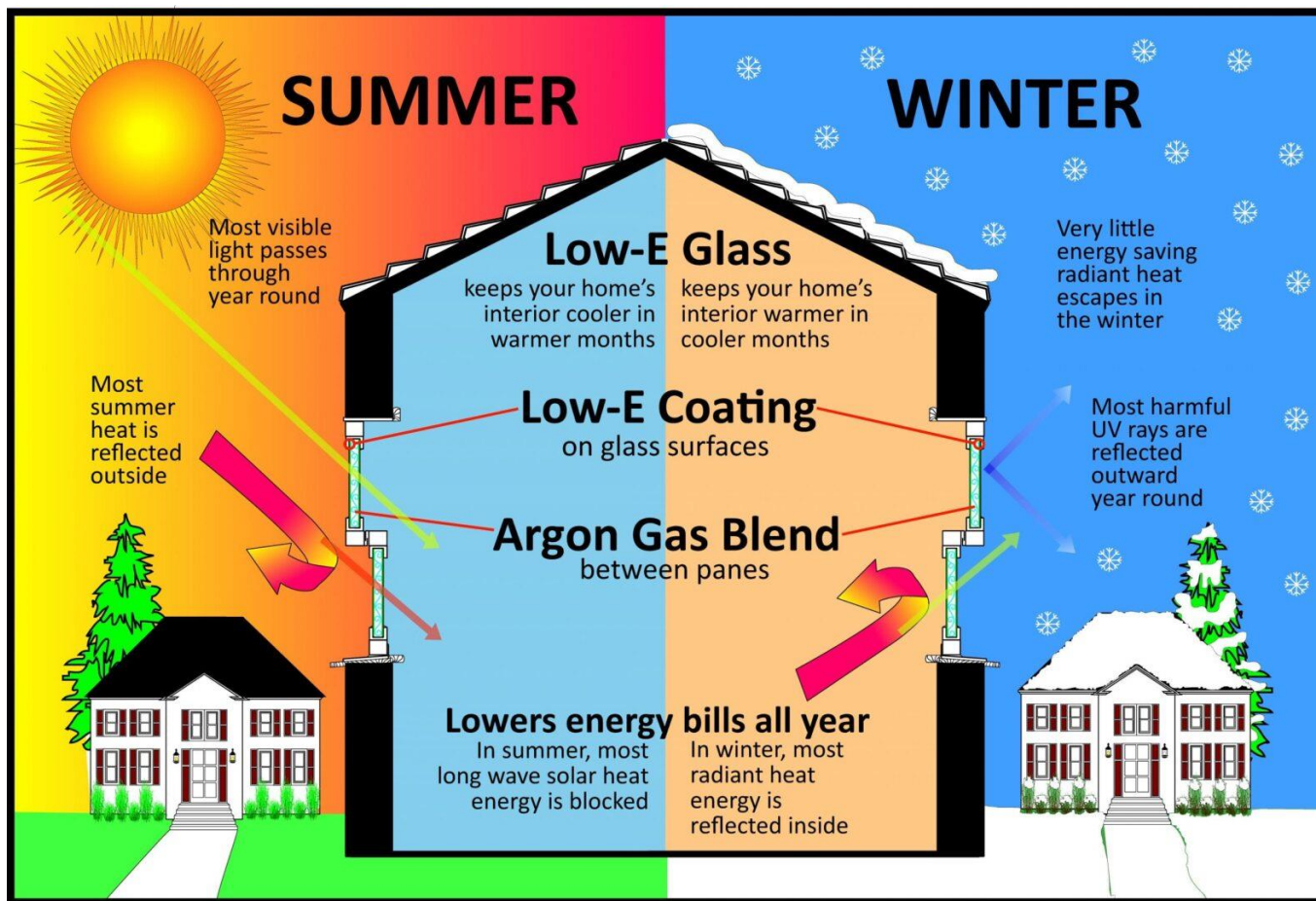


**most surfaces are opaque to long-wave radiation therefore
absorptivity typically = emissivity**



Low emissivity = reflects much of the radiant thermal energy

Low-e glass as one practical example of a low emissivity surface



Is long-wave reflection important?

Energy conservation	Transmissivity		Absorptivity		Reflectivity	
	$\psi_{o,LW}$	+	$\zeta_{o,LW}$	+	$\alpha_{o,LW}$	= 1.0

★

However, most surfaces are opaque to long-wave (i.e., $\psi_{o,LW} \sim 0$) so

$$\zeta_{o,LW} + \alpha_{o,LW} = 1.0$$

and since $\zeta_{o,LW} = \epsilon_{o,LW}$ (according to Kirchhoff's Law)

$$\alpha_{o,LW} =$$

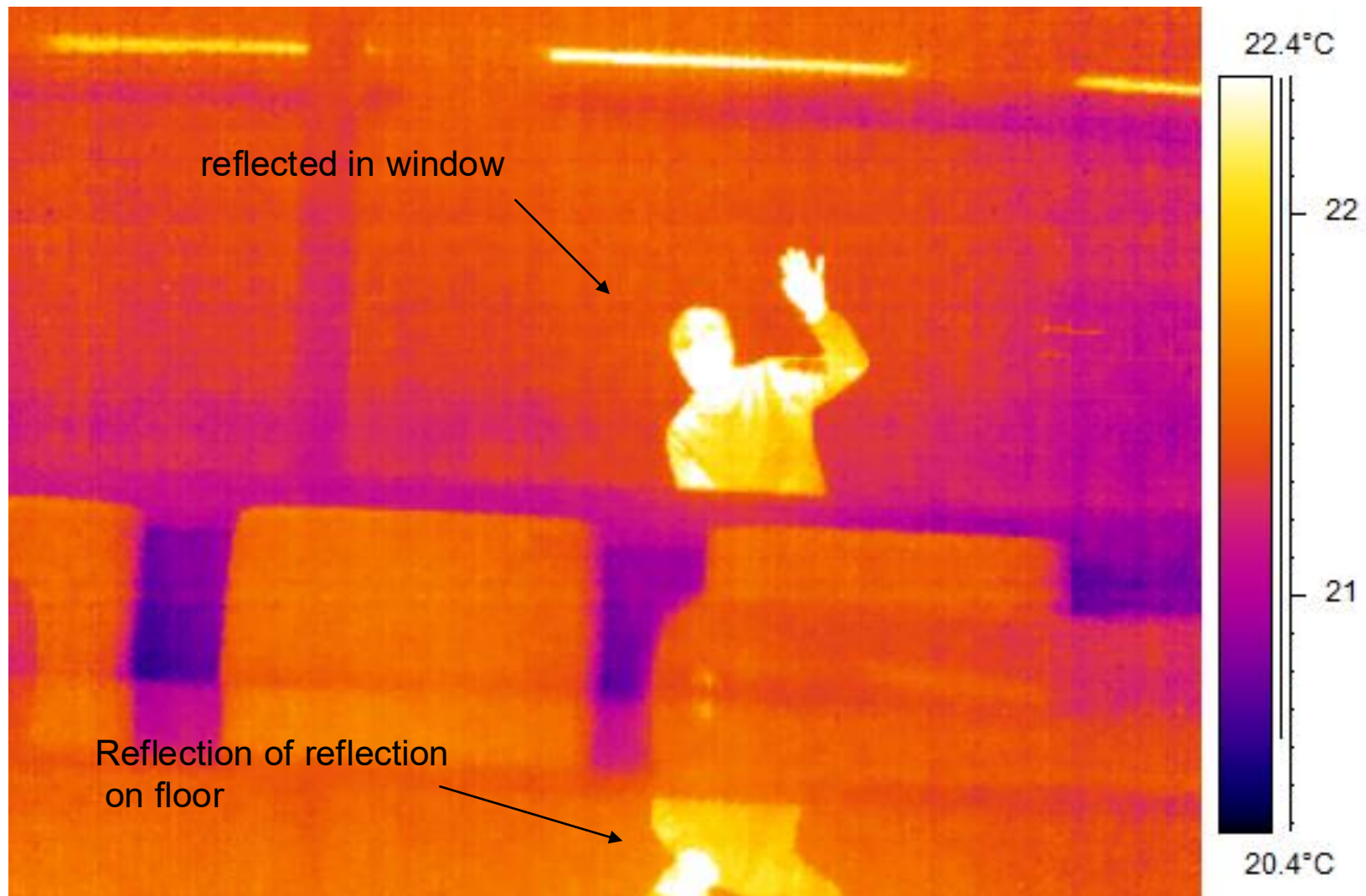
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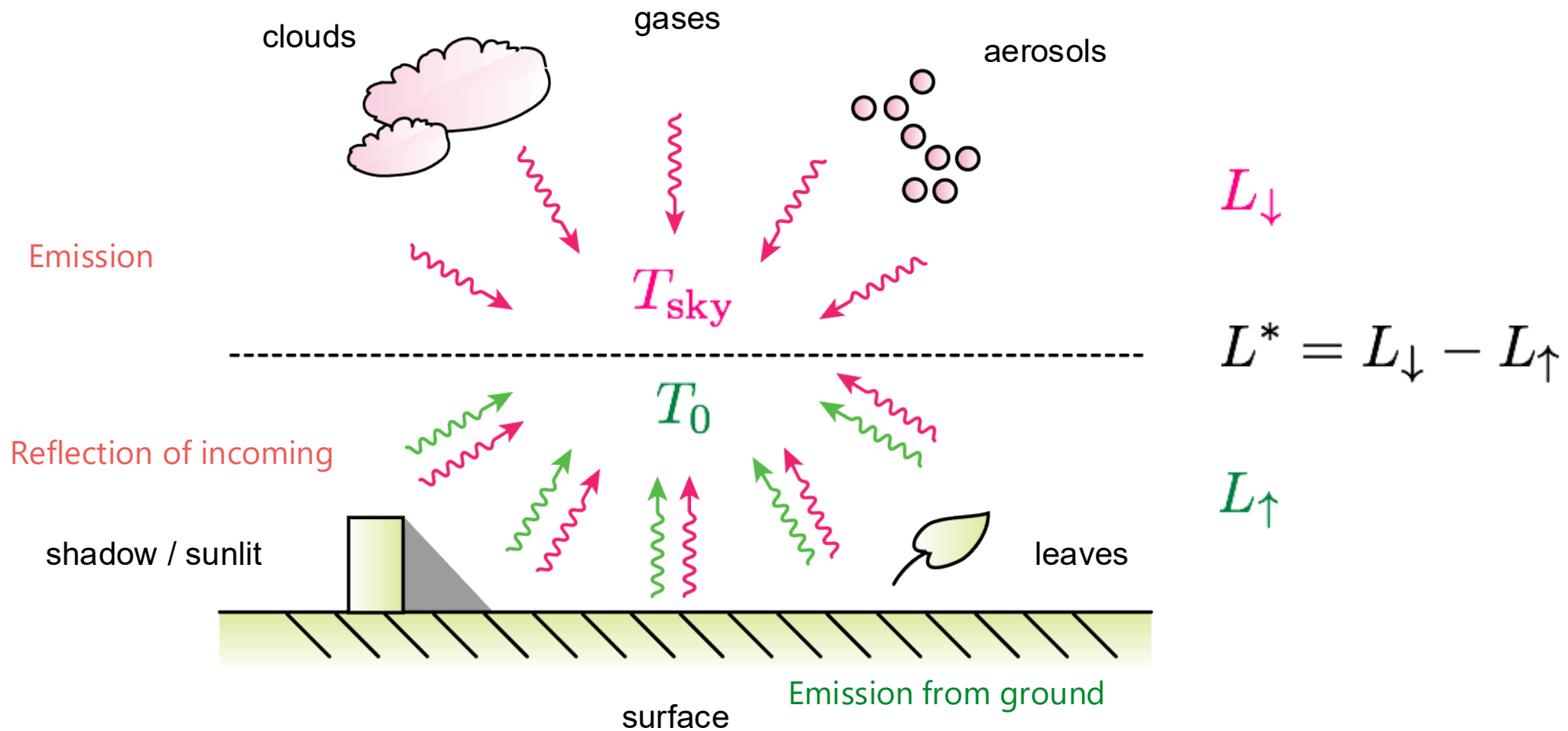


The subscript '0' refers to values of the land surface (in contrast to atmosphere with subscript a). In the upcoming slides we will implicitly assume values are in the long-wave band, and omit the subscript 'LW'

Long-wave reflection



Long-wave radiation exchange of a surface

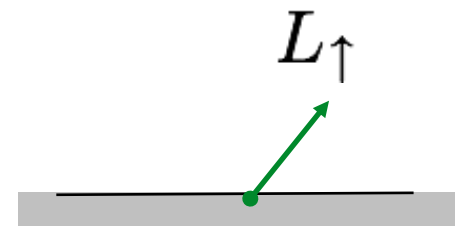


Calculating L_{\uparrow}

$$\varepsilon = \frac{\text{radiative flux density emitted by a body}}{\text{radiative flux density emitted by a blackbody}}$$

defines blackbody as $\varepsilon = 1.0$, and grey bodies as $\varepsilon < 1.0$.

$$L_{\uparrow} = \overset{\text{emission}}{\varepsilon_o \sigma T_o^4}$$

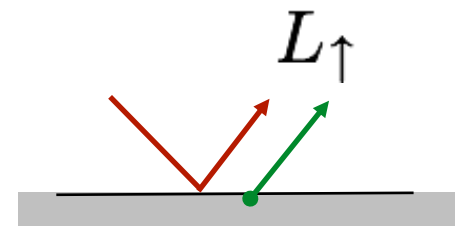


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$$L_{\uparrow} = \overset{\text{emission}}{\varepsilon_o \sigma T_o^4} + \underbrace{(1 - \zeta_o)}_{\alpha_o} L_{\downarrow}$$

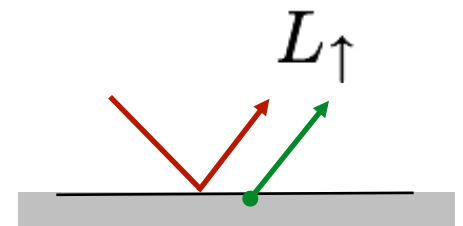


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where ζ_o - absorptivity of surface in the long-wave, α_o reflectivity of surface in the long-wave and from Kirchhoff's law ($\zeta_{\lambda} = \varepsilon_{\lambda}$):

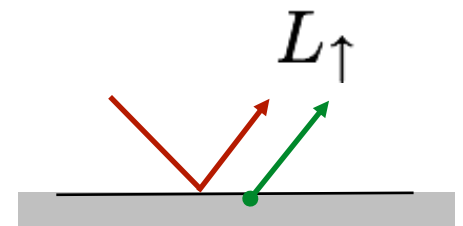
where subscript '0' indicates surface value.

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$$L_{\uparrow} = \varepsilon_o \sigma T_o^4 + (1 - \varepsilon_o) L_{\downarrow} \quad \star$$

where subscript '0' indicates surface value.

Net long-wave radiation flux density L^*

The long-wave net radiation L^* at the surface is the difference between the input from above L_{\downarrow} and the output from emission and reflected L_{\uparrow} :

$$L^* = L_{\downarrow} - L_{\uparrow}$$

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$$L^* = L_{\downarrow} - (\epsilon_o \sigma T_o^4 + \overbrace{(1 - \zeta_o)}^{\alpha_o} L_{\downarrow})$$

Net long-wave radiation flux density L^*

The long-wave net radiation L^* at the surface is the difference between the input from above L_{\downarrow} and the output from emission and reflected L_{\uparrow} :

$$L^* = L_{\downarrow} - L_{\uparrow}$$

$$L^* = L_{\downarrow} - (\underbrace{\epsilon_o \sigma T_o^4}_{\text{emitted}} + \underbrace{\overbrace{(1 - \zeta_o)}^{\alpha_o} L_{\downarrow}}_{\text{reflected}})$$

Kirchhoff's Law

$$L^* = L_{\downarrow} - (\epsilon_o \sigma T_o^4 + (1 - \epsilon_o) L_{\downarrow})$$

Question for discussion

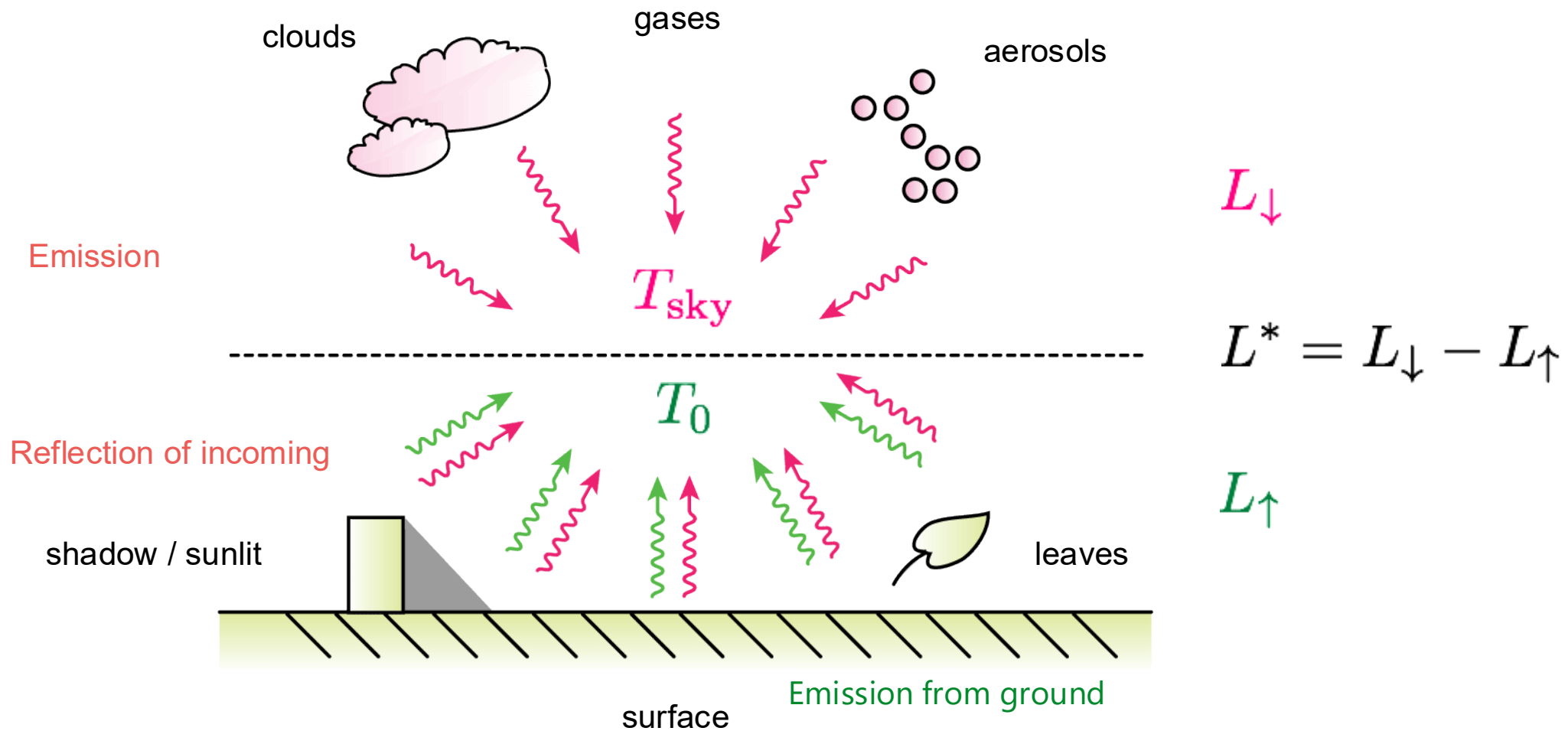
What influences L_{\downarrow} ?

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Long-wave radiation exchange of a surface



Long-wave radiation - example of measurements

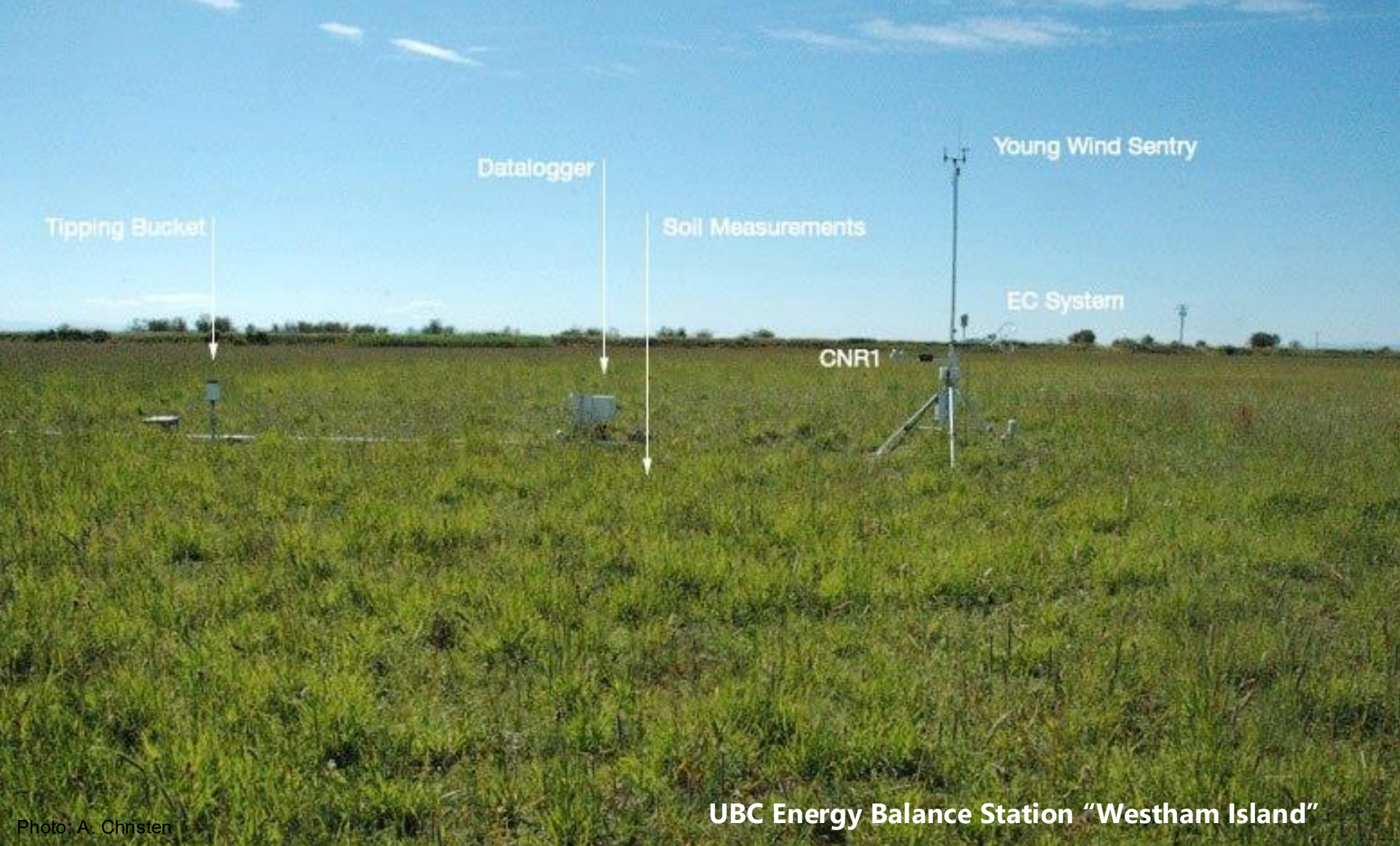


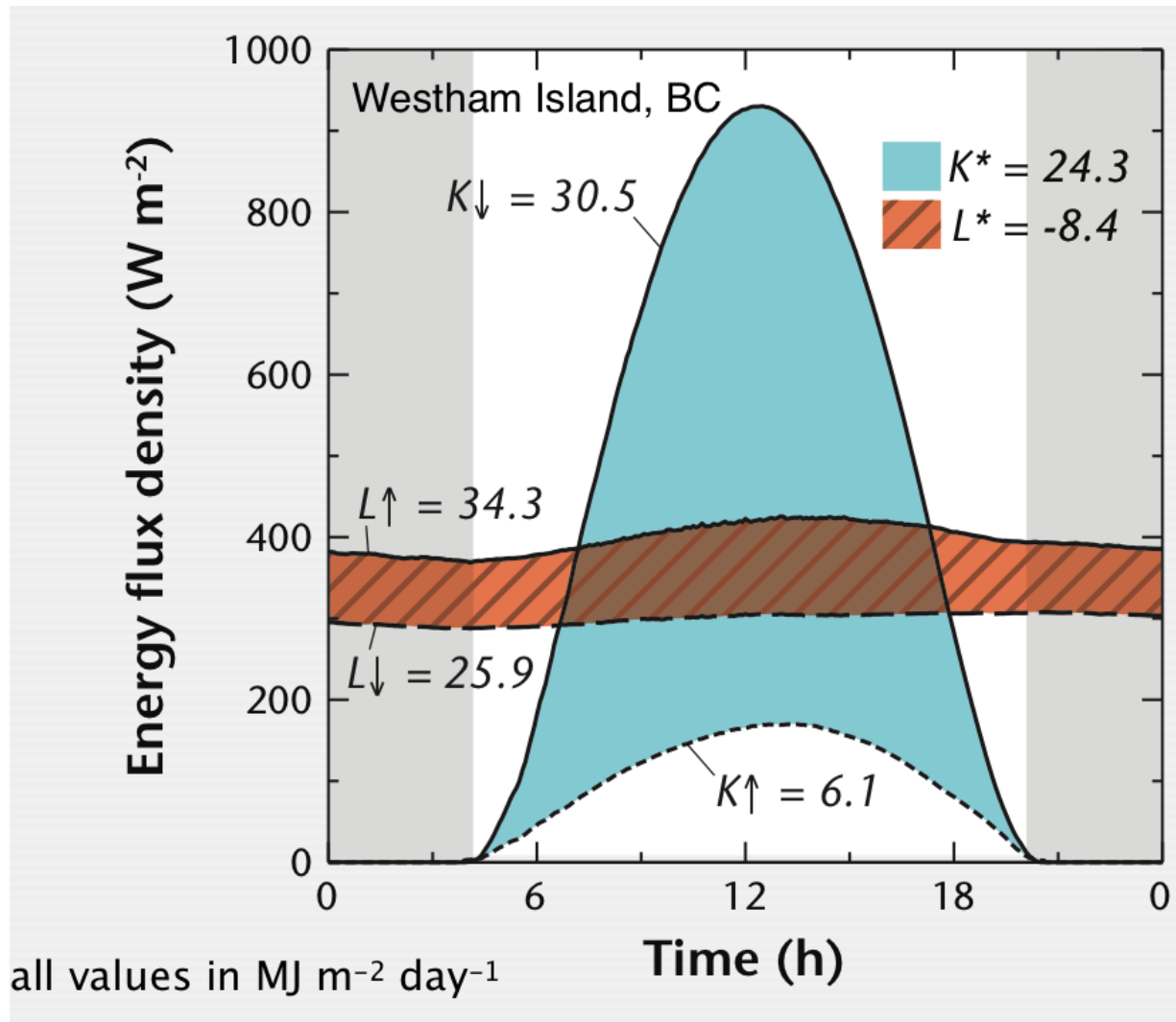
Photo: A. Christen

UBC Energy Balance Station "Westham Island"

Long-wave radiation - example of measurements



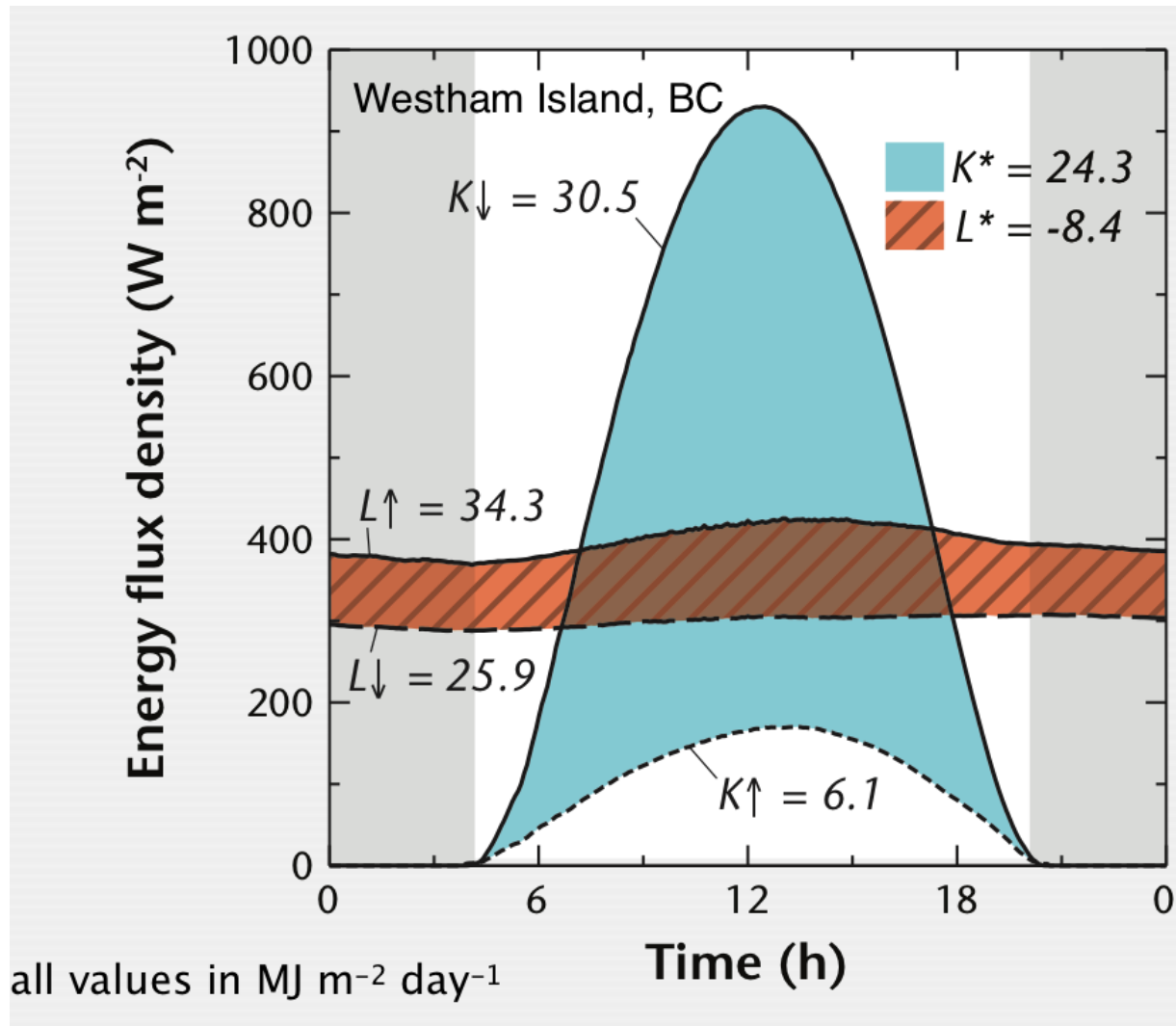
Diurnal course of radiation - clear sky day



Images from Katkam in Vancouver

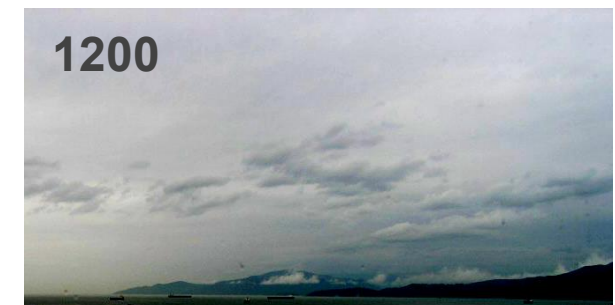
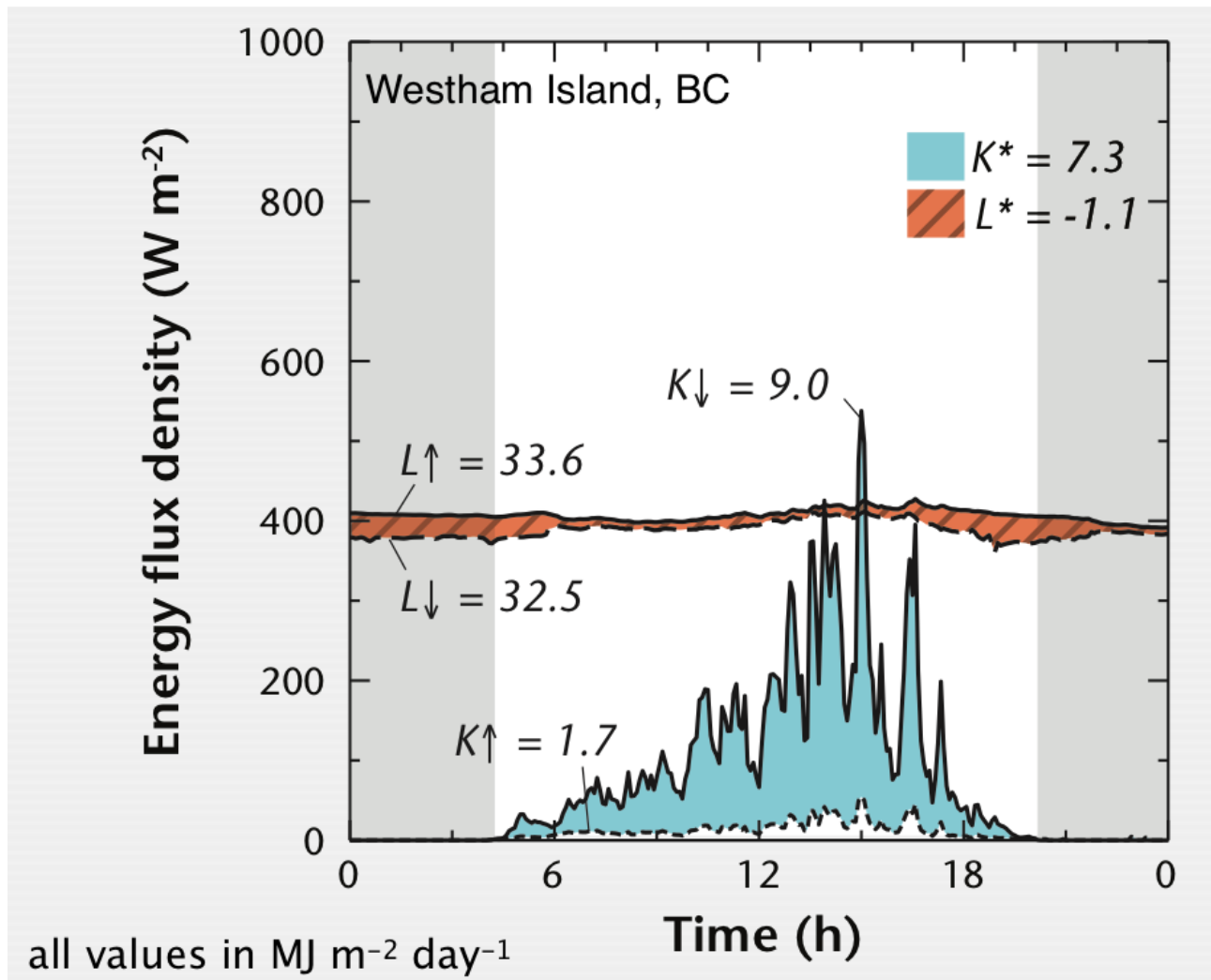


Class activity - Why is $L_{up} > L_{down}$?



Images from Katkam in Vancouver

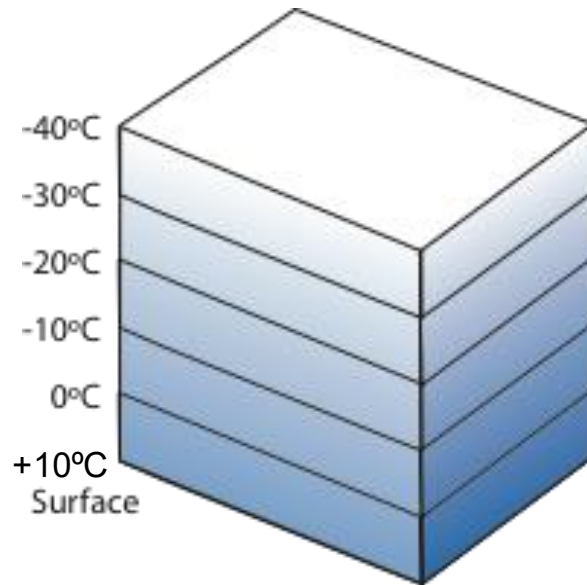
Diurnal course of radiation - overcast / broken sky day



Images from Katkam in Vancouver

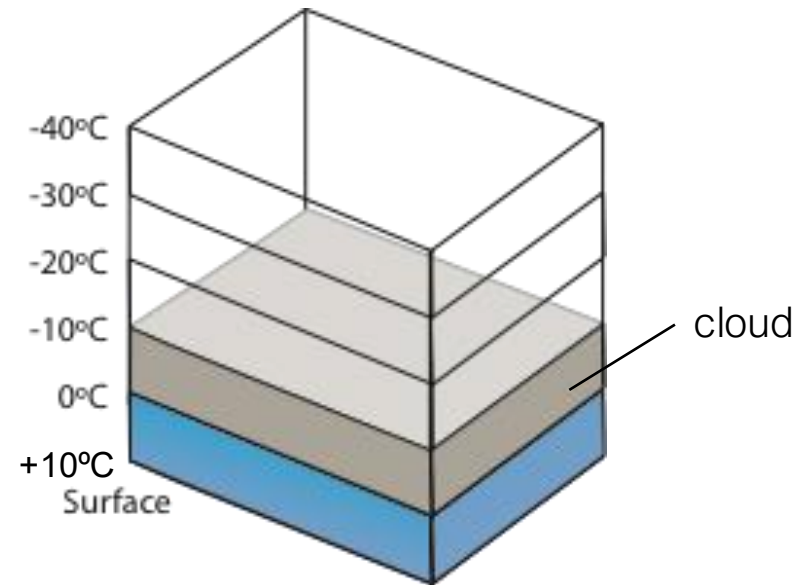
Effect of clouds on longwave irradiance

With a clear-sky, L_{\downarrow} originates from **all layers** of the atmosphere, because the atmosphere is **partly transparent** ('atmospheric window' open)



Signal is a mixture from -40°C to +10°C

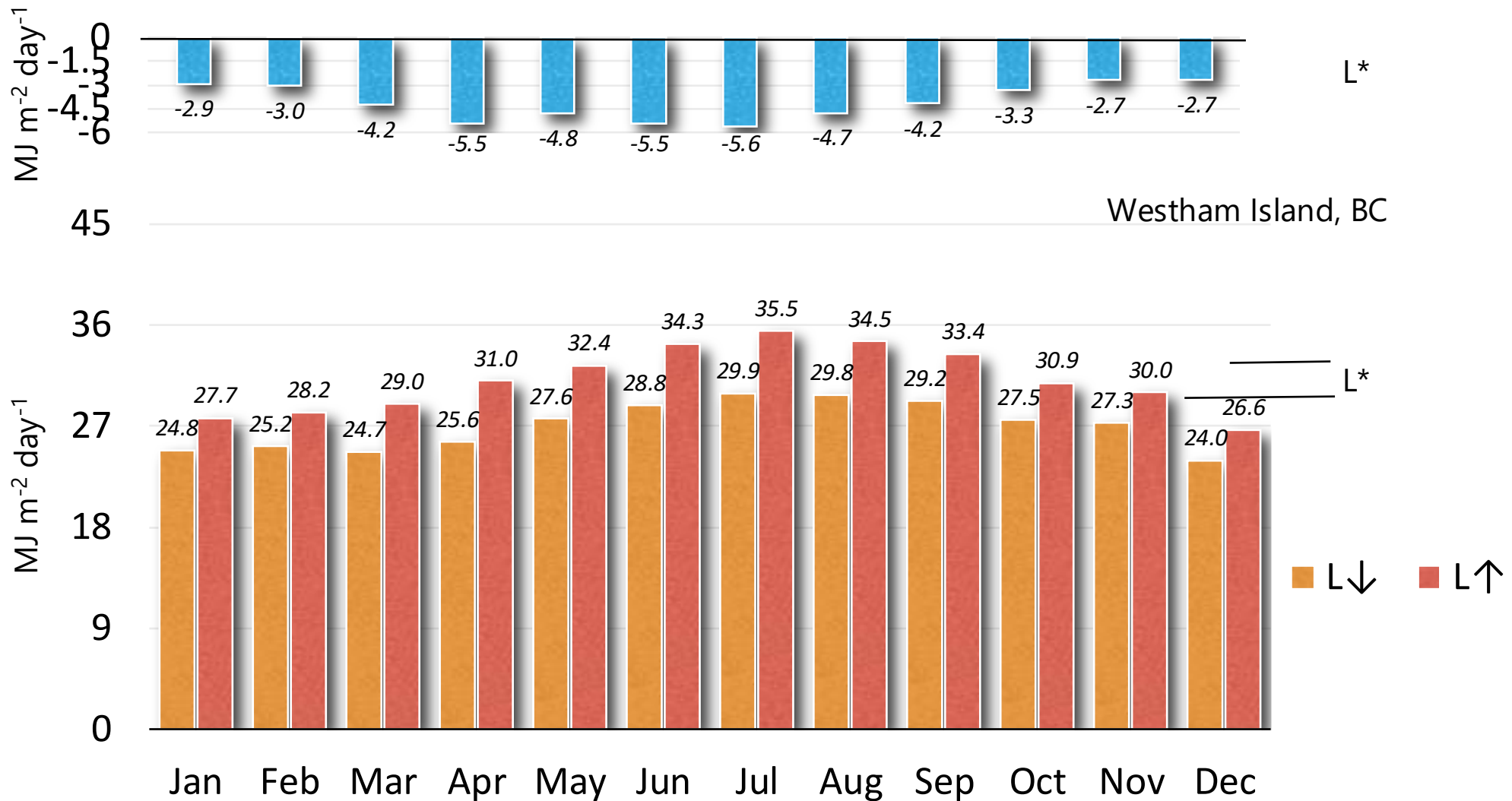
With clouds L_{\downarrow} originates from **the cloud base** and the atmosphere below the cloud, because the cloud is **opaque to long wave** ('atmospheric window' closed)



Signal is a mixture from 0° to +10°C

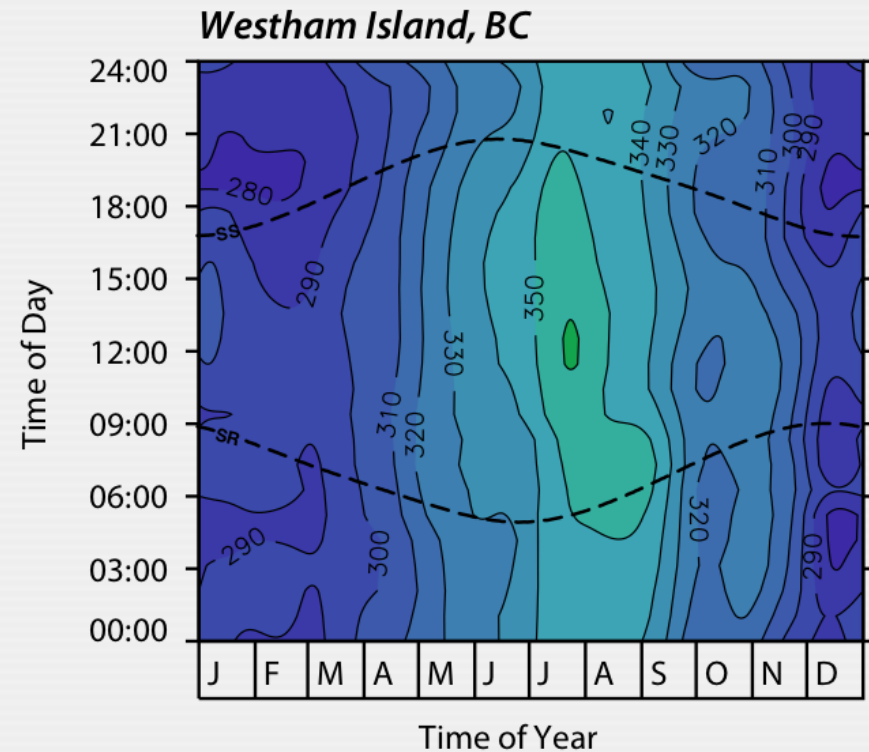
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Monthly totals of L_{\downarrow} , L_{\uparrow} and L^* in Vancouver

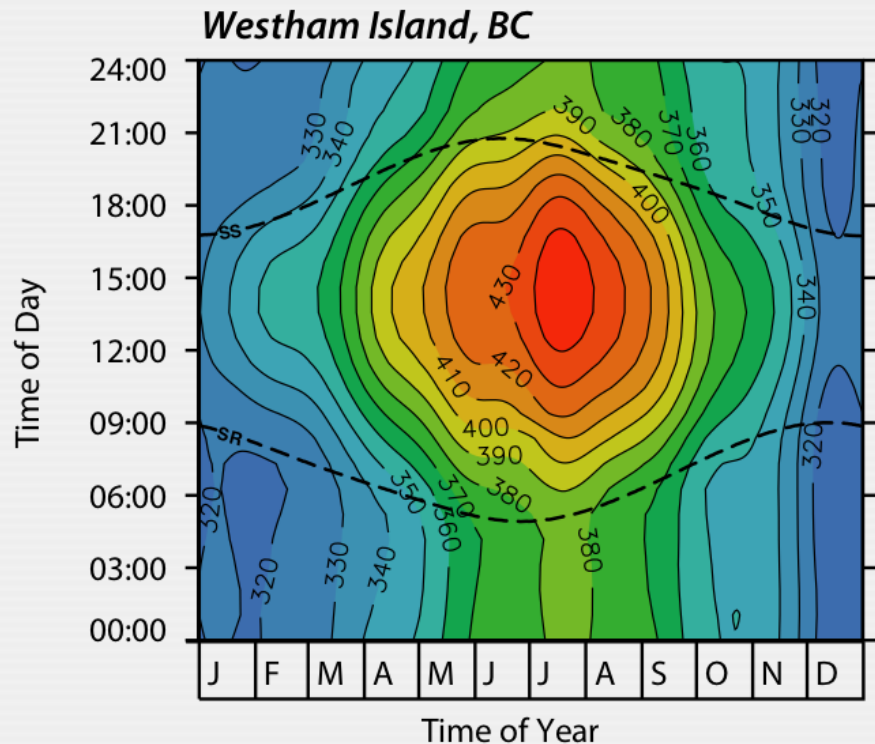


'Fingerprint' of L_{\downarrow} and L_{\uparrow} measured in Vancouver

(a) L_{\downarrow}



(b) L_{\uparrow}



Take home points

- Radiation laws apply the same way to the longwave part of the spectrum - but **Kirchhoff's law** and the concept of **emissivity** become relevant.
- The net-long wave radiation is driven by the difference in apparent sky and surface temperatures and hence **clouds and thermal surface properties** are controlling radiative exchange in the long-wave.